

RTM Nagpur University
Syllabus (Theory)

Semester	Course Title (Subject)	Hours / Week			Cr edi ts	Maximum Marks			Exam Durati on (Hrs.)
		L	T	P		Contin ual Assess ment	Unive rsity Exam inatio n	Total	
B.E. III Sem	Engineering Thermodynamics	3	1	-	4	30	70	100	03

Sr. No.	Course Objective The objective of this course is–
1	This course deals with the fundamentals of Thermodynamics, including thermodynamic systems and properties, relationships among the thermos-physical properties, the laws of thermodynamics and applications of these fundamental laws in thermodynamic systems
2	To present a comprehensive and rigorous treatment of classical thermodynamics while retaining an engineering perspective.
3	Explain the working principle of various power cycles used in thermal systems.
Course Outcomes	
After successful completion of this course, the student will be able to:	
CO1	Explain thermodynamics concepts, relate laws of the ideal gas, identify various thermodynamic processes and apply the laws to determine the energy transfer in terms of heat and work.
CO2	Explain the first law of thermodynamics and apply the law to evaluate open, closed systems, thermal components and devices.
CO3	Interpret the second law of thermodynamics, entropy, and apply the law to evaluate heat engine, heat pump, and refrigerator performance.
CO4	Relate various steam properties, and analyze the different types of processes using steam as working fluid to determine the energy transfer in terms of heat and work.
CO5	Compare various power cycles and analyze the cycles to determine the energy transfer in terms of heat, work and efficiency.

SYLLABUS	
Contents	No of hours
Unit I Basic concepts of Thermodynamics, Systems and their types, Property, State, Process, Phase, Cycles. Comparison of microscopic and macroscopic approaches. Path and point functions. Thermodynamic Equilibrium. Zeroth law of thermodynamics and its significance for temperature measurement Introduction to First law of thermodynamics, Energy transfer, Heat and work transfer.	10

<p>Ideal Gas laws: Boyle's law, Charle's law, Gay-Lussac's law, Avagadro's law, Equation of state, General gas equation, Specific Heat, Universal gas constant.</p> <p>Thermodynamic Processes: Constant pressure, Constant volume, Isothermal, Isentropic and Polytropic process, representation on P-V and T-s Diagram, Calculation of Heat transfer, Work done, Change in Internal Energy and Enthalpy for these processes.</p>	
<p>Unit II</p> <p>The first law of Thermodynamics for Closed System undergoing a process and cycle (Control Mass System) and Open System (Control Volume System)</p> <p>Steady Flow process applies to Compressor, Pump, Turbine, Boiler, Steam Nozzle, Throttling Device, Heat Exchanger, Fan and blower.</p> <p>(Analytical treatment on First law applied to thermodynamic processes and cycles and Steady low energy equation applied to various flow devices is expected).</p>	9
<p>Unit III</p> <p>Second Law of Thermodynamics:- Heat Reservoir, source and sink. Heat Engine, Refrigerator, Heat Pump, Kelvin-Plank and Clausius Statements, Perpetual Motion Machine I and II, Carnot Cycle, Thermodynamic Temperature scale.</p> <p>Entropy:- Clausius Inequality, Entropy, Principle of Increase of Entropy, Change in Entropy for different thermodynamics processes with T-S Diagram, Reversible and Irreversible Processes. (Simple analytical treatment on COP calculation is expected)</p>	9
<p>Unit IV</p> <p>Properties of Steam:- Formation of steam and its thermodynamic properties like Sensible Heat, Latent Heat, Critical State, Triple Point, Wet Steam, Dry Steam, Superheated Steam, Dryness Fraction, Enthalpy, Internal Energy of Steam, External Work Done during Evaporation, T-S Diagram, Mollier Chart, Work and Heat Transfer during various Thermodynamic Processes with steam as working fluid. Measurement of Dryness Fraction using various Calorimeters. (Analytical Treatment using steam table and Mollier chart is expected)</p>	9
<p>Unit V</p> <p>Power Cycles: - Otto Cycle, Diesel Cycle, Dual Cycle, Brayton Cycle, Representation on P-v and T-s diagrams. The equation for work done, heat transfer, air standard efficiency, and mean effective pressure. Comparison of Otto, Diesel and Dual cycles. Introduction to simple vapour power cycle, i.e., Rankine cycle (Analytical treatment in terms of calculation Work done & efficiency analysis is expected on Otto Cycle, Diesel Cycle and Dual Cycle)</p>	9
Total Hours	46

Sr. No.	List of Tutorials
01	Application of first law to control mass (closed system) system
02	Application of first law to control volume (open system) system
03	Determination of Heat transfer, Work done, Change in Internal Energy and Enthalpy of

	various thermodynamic processes and cycles.
04	Determination of various properties of steam by using Steam table and Mollier chart
05	Application of second law to heat engine, refrigerator and heat pump.
06	Thermodynamic analysis of Otto cycle.
07	Thermodynamic analysis of Diesel cycle.
08	Thermodynamic analysis of Dual cycle and Brayton cycle.

References:

Text Books Recommended:

1. Engineering Thermodynamics, P. K. Nag, Tata McGraw-Hill Publications
2. Thermodynamics, S. C. Gupta, Pearson Publications
3. Thermal Engineering, P. L. Ballani, Khanna Publications
4. Engineering Thermodynamics, S.S. Khandare, Charotar Publication House
5. Engineering Thermodynamics, R. K. Rajput, Laxmi Publication

Reference Books Recommended:

1. Thermodynamics and Engineering approach, Yunus A. Cengel, Michael A. Boles, Tata McGraw-Hill Publications
2. Engineering Thermodynamics, D. P. Mishra, Cengage Learning Publications
3. Engineering Thermodynamics, Gordon Rogers, Pearson Publications